

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t902189982>

CONSTRUCTION OF THE PYRAZOLO[1,5-*a*][6,1]BENZODIAZONINE AND PYRAZOLO[1,5-*a*][7,1]BENZODIAZECINE SKELETONS

Gianluigi Broggin^a; Luisa Garanti^a; Giorgio Molteni^a; Gaetano Zecchi^a

^a Dipartimento di Chimica Organica e Industriale and Centro C.N.R., Milano, Italy

To cite this Article Broggin, Gianluigi, Garanti, Luisa, Molteni, Giorgio and Zecchi, Gaetano (1996) 'CONSTRUCTION OF THE PYRAZOLO[1,5-*a*][6,1]BENZODIAZONINE AND PYRAZOLO[1,5-*a*][7,1]BENZODIAZECINE SKELETONS', *Organic Preparations and Procedures International*, 28: 6, 699 – 701

To link to this Article: DOI: 10.1080/00304949609356735

URL: <http://dx.doi.org/10.1080/00304949609356735>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

4. a) H. Wamhoff, W. Schupp, A. Kirfel and G. Will, *J. Org. Chem.*, **51**, 149 (1986); b) H. Wamhoff and W. Schupp, *ibid.*, **51**, 2787 (1986); c) E. B. Walsh and H. Wamhoff, *Chem. Ber.*, **122**, 1673 (1989); d) H. Wamhoff, A. Schmidt and M. Nieger, *Tetrahedron Lett.*, **32**, 4473 (1991); e) P. Molina and M. J. Vilaplana, *Synthesis*, 474 (1990); f) P. Molina, A. Lorenzo and E. Aller, *ibid.*, 297 (1992).
5. P. Molina and M. J. Vilaplana, *ibid.*, 1197 (1994).
6. W. Bunge, "Houben-Weyl Methoden der Organische Chemie", Vol. I/2, pp. 765-868.

CONSTRUCTION OF THE PYRAZOLO[1,5-*a*][6,1]BENZODIAZONINE AND PYRAZOLO[1,5-*a*][7,1]BENZODIAZECINE SKELETONS

Submitted by
(02/05/96)

Gianluigi Brogгинi, Luisa Garanti*, Giorgio Molteni and Gaetano Zecchi

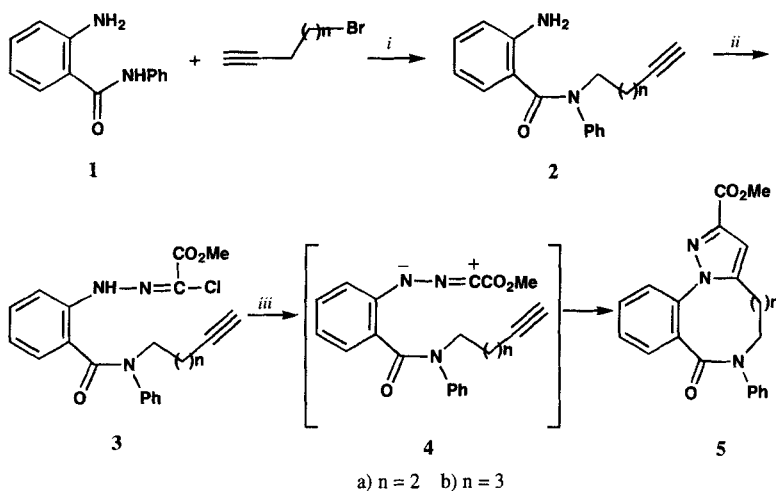
*Dipartimento di Chimica Organica e Industriale and Centro C.N.R.
Via Golgi 19, 20133 Milano, Italy*

Intramolecular 1,3-dipolar cycloadditions offer a synthetic entry to a large variety of heterocyclic systems containing a five-membered ring fused to another ring of variable size.¹ The latter parameter, however, markedly affects the entropy aspect and thus the effectiveness of the intramolecular process.² We now report the successful use of intramolecular nitrilimine cycloaddition to construct the hitherto unreported pyrazolo[1,5-*a*][6,1]benzodiazonine and pyrazolo[1,5-*a*][7,1]benzodiazecine systems.

The synthetic sequence involves the following stages: (i) site-selective alkynylation of the *o*-aminobenzanilide (**1**);⁴ (ii) diazotisation of **2** and subsequent coupling with methyl α -chloroacetate to give the hydrazoneyl chlorides **3**; (iii) treatment of the latter with silver carbonate in order to generate the transient nitrilimines **4**, whose intramolecular cycloaddition led to the target system **5**. In view of the well-known factors working against the formation of large rings,³ the cyclization yields of **4** are highly satisfactory thus making the present synthesis worthy of attention.

EXPERIMENTAL SECTION

Melting points were determined on a Büchi apparatus and are not corrected. IR spectra were recorded on a FT IR Perkin Elmer 1725 X spectrophotometer. Mass spectra were taken with a WG-70EQ apparatus. ¹H NMR spectra were obtained on a Bruker 300 MHz apparatus, chemical shifts are given as ppm from TMS. Compound **1** was prepared according to the literature.⁴



i) K_2CO_3 , NaOH, $n\text{-Bu}_4\text{N}^+ \text{HSO}_4^-$, benzene, 70° ii) 1) NaNO_2 , HCl, AcOH 2) $\text{MeCOCHClCO}_2\text{Me}$

iii) Ag_2CO_3 , dioxane, Δ

Alkyne Anthranilates 2. Typical Procedure.- A solution of **1** (18 mmol) in dry benzene (80 mL) was treated with K_2CO_3 (20 mmol), NaOH (68 mmol) and $n\text{-Bu}_4\text{N}^+ \text{HSO}_4^-$ (2 mmol) and heated to 70° . Then the appropriate alkyne bromide (25 mmol) in dry benzene (12 mL) was added dropwise (20 min) at 70° . The mixture was refluxed for 8 hrs, then washed with water (250 mL) and dried (Na_2SO_4). The solvent was removed under reduced pressure and the residue was chromatographed on a silica gel column with light petroleum-ethyl acetate 2:1 as eluent to give **2**.

Compd 2a: 2.00 g (40%); thick oil; IR (nujol): 3475, 3360, 3280, 1640 (cm^{-1}); $^1\text{H NMR}$ (CDCl_3): δ 1.92 (t, $J = 2.5$, 1H), 2.10-2.45 (m, 4H), 3.95 (t, $J = 7.0$, 2H), 4.60 (br s, 2H), 6.12-7.70 (m, 9H); MS m/z 278 (M^+).

Anal. Calcd. for $\text{C}_{18}\text{H}_{18}\text{N}_2\text{O}$: C, 77.66; H, 6.52; N, 10.07. Found: C, 77.53; H, 6.47; N, 9.95

Compd 2b: 1.84 g (35%); thick oil; IR (nujol): 3460, 3360, 3300, 1645 (cm^{-1}); $^1\text{H NMR}$ (CDCl_3): δ 1.55-1.74 (m, 4H), 1.91 (t, $J = 2.5$, 1H), 2.21 (dt, $J = 7.0, 2.5$, 2H), 3.95 (t, $J = 7.0$, 2H), 4.60 (br s, 2H), 6.20-7.60 (m, 9H); MS m/z 292 (M^+).

Anal. Calcd. for $\text{C}_{19}\text{H}_{20}\text{N}_2\text{O}$: C, 78.04; H, 6.90; N, 9.59. Found: C, 77.91; H, 6.89; N, 9.70

Hydrazonyl Chlorides 3. Typical Procedure.- Sodium nitrite (13 mmol) was added portionwise to a solution of the alkyne anthranilate **2** (6 mmol) in 1N aqueous HCl (30 mL) and glacial acetic acid (5 mL) with vigorous stirring and cooling at 0°C . After 30 min, the cold mixture was treated with sodium acetate to adjust the pH to 5 and then methyl 2-chloroacetoacetate (6 mmol) was added with cooling and stirring for 4hrs. The mixture was extracted with Et_2O (50 mL) and the organic layer was washed with 5% aqueous NaHCO_3 until neutral, dried and evaporated. Crystallization from diisopropyl ether gave the hydrazonyl chlorides **3**.

Compd 3a: 1.43 g (60%); mp. 97° ; IR (nujol): 3250, 1740, 1690 (cm^{-1}); $^1\text{H NMR}$ (CDCl_3): δ 1.93 (t, $J = 2.5$, 1H), 2.10-2.35 (m, 4H), 3.90 (s, 3H), 4.02 (t, $J = 8.0$, 2H), 6.55-7.65 (m, 9H), 10.45 (br s, 1H);

MS m/z 397 (M^+).

Anal. Calcd. for $C_{21}H_{20}ClN_3O_3$: C, 63.46; H, 5.08; Cl, 8.81; N, 10.58

Found: C, 63.30; H, 4.92; Cl, 8.88; N, 10.64

Compd **3b**: 1.85 g (75%); mp. 89°; IR (nujol): 3290, 1760, 1640 (cm^{-1}); 1H NMR ($CDCl_3$): δ 1.55-1.73 (m, 4H), 1.91 (t, $J = 2.5$, 1H), 2.22 (dt, $J = 7.0, 2.5$, 2H), 3.94 (s, 3H), 3.97 (t, $J = 7.0$, 2H), 6.55-7.61 (m, 9H), 10.45 (br s, 1H); MS m/z 411 (M^+).

Anal. Calcd. for $C_{22}H_{22}ClN_3O_3$: C, 64.21; H, 5.39; Cl, 8.51; N, 10.22

Found: C, 64.32; H, 5.45; Cl, 8.60; N, 10.35.

Reaction of Hydrazone Chlorides 3 with Silver Carbonate. Typical Procedure.- A solution of the hydrazone chloride **3** (3 mmol) in dry dioxane (300 mL) was treated with silver carbonate (14 mmol) and the mixture was refluxed *in the dark* for 12 hrs. The undissolved material was filtered off and the filtrate was evaporated under reduced pressure. The residue was chromatographed on a silica gel with dichloromethane-diethyl ether 1:1 as eluent to give **5**.

Compd **5a**: 0.51 g (47%); mp. 197° (from hexane-benzene); IR (nujol): 1730, 1720 (cm^{-1}); 1H NMR ($CDCl_3$): δ 1.50-1.82 (m, 4H), 2.35-2.45 (m, 1H), 3.00-3.08 (m, 1H), 3.90 (s, 3H), 6.78 (s, 1H), 6.80-7.63 (m, 9H); MS m/z 361 (M^+).

Anal. Calcd. for $C_{21}H_{19}N_3O_3$: C, 69.78; H, 5.30; N, 11.63. Found: C, 69.65; H, 5.25; N, 11.71

Compd **5b**: 0.48 g (43%); mp. 203° (from hexane-benzene); IR (nujol): 1730, 1720 (cm^{-1}); 1H NMR ($CDCl_3$): δ 1.20-2.30 (m, 6H), 2.90 (br t, $J = 7.5$, 2H), 3.92 (s, 3H), 6.78 (s, 1H), 6.80-7.60 (m, 9H); MS m/z 375 (M^+).

Anal. Calcd. for $C_{22}H_{21}N_3O_3$: C, 70.37; H, 5.64; N, 11.20. Found: C, 70.41; H, 5.73; N, 11.36

Acknowledgements.- We are grateful to MURST and CNR for financial support.

REFERENCES

1. A. Padwa, "1,3-Dipolar Cycloaddition Chemistry", Vol. 2, p. 277-406, A. Padwa, ed.; Wiley-Interscience, New York, NY, 1984; "Advances in Cycloaddition", Vol. 2, p. 1-89, D. P. Curran, ed.; JAI Press, London, 1990.
2. a) H. Meier, H. Heimgartner and H. Schmid, *Helv. Chim. Acta*, **60**, 1087 (1977); b) H. Meier and H. Heimgartner, *ibid.*, **69**, 927 (1986); c) J. Brokatzky Geiger and W. Eberbach, *Heterocycles*, **20**, 1519 (1983); d) J. Brokatzky-Geiger and W. Eberbach, *Chem. Ber.*, **117**, 2157 (1984); e) W. Eberbach, I. Heinze, K. Knoll, H. Fritz and F. Borke, *Helv. Chim. Acta*, **71**, 404 (1988); f) I. Heinze, K. Knoll, R. Müller and W. Eberbach, *Chem. Ber.*, **122**, 2147 (1989); (g) G. Broggin, L. Garanti, G. Molteni and G. Zecchi, *J. Chem. Res.*, 385 (S), 2389 (M), (1995).
3. K. Ziegler, In "Methoden der Organischen Chemie (Houben-Weyl)", E. Müller, ed.; Vol. 4/2, p. 729-822, Georg Thieme Verlag, Stuttgart, 1955; M. A. Winnik, *Chem. Rev.*, **81**, 491 (1981); G. Illuminati and L. Mandolini, *Acc. Chem. Res.*, **14**, 95 (1981).
4. W. E. Coyne and J. W. Cusic, *J. Med. Chem.*, **11**, 1208 (1968).